

Determine the oxidation number of the underlined elements in the following compounds:

<u>O</u> F ₂	K <u>N</u> O ₃	K ₂ <u>Cr</u> ₂ O ₇	Al ₂ (<u>S</u> O ₄) ₃
<u>N</u> O ₃ ⁻	<u>C</u> O ₃ ²⁻	<u>P</u> O ₄ ³⁻	<u>N</u> H ₄ ⁺
	<u>Mn</u> O ₄ ⁻	<u>M</u> g	