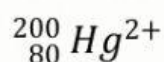


## CHEMISTRY NOTES & EXERCISE

**Complete the following sentence:**

Isotopes are atoms of same element with the same \_\_\_\_\_ but different \_\_\_\_\_

Look at the isotope notation below:



**Choose the correct number of sub-atomic particles:**

Proton

Neutron

Electron

**Choose the correct statement about carbon**

Component of graphite

Consist of only one isotope

It is positively charge

**Which of the following is correct about neutron?**

It is neutral

It is in the nucleus

It is positively charge



**Match the following subatomic particle and its charge**

Proton	Negative
Neutron	Positive
Electron	Neutral

**Drag the answers to the correct place**

Negative      Positive      Neutral

Subatomic particle	Charge
Proton	
Neutron	
Electron	

