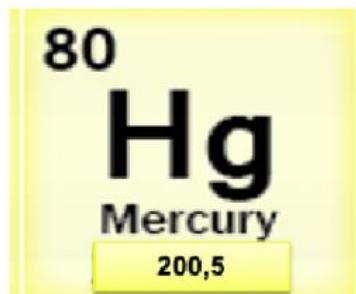
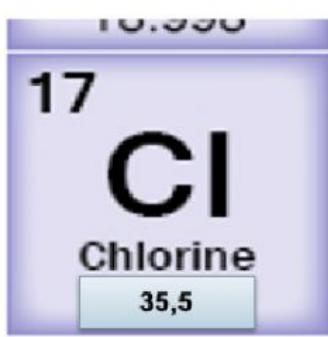
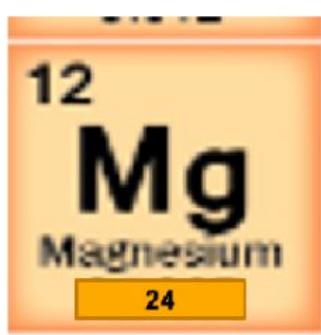


# Practice

-Complete the PEN of each element below.



Proton :

Electron :

Neutron :

Proton :

Electron :

Neutron :

Proton :

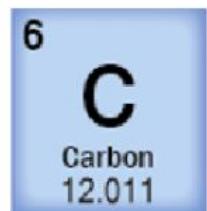
Electron :

Neutron :

# Summarize

1. What is the atomic mass of the element shown?

- a.) 18
- b.) 13
- c.) 12
- d.) 11



2.) How many electrons would there be in a neutral atom of the element above?

- a.) 6
- b.) 12
- c.) 6.011
- d.) 18

3.) T/F: The periodic table is arranged by atomic mass

4.) T/F: Most of the mass of an element is in the electron shell.