

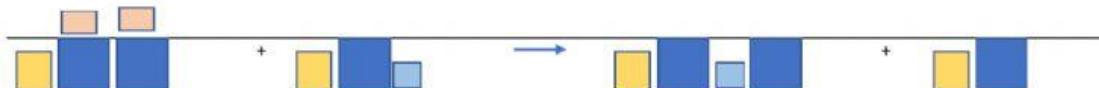
Fill in the box with the correct chemical formula and number to complete the balance chemical equation

e.g: Sodium burn in oxygen



$\text{Na} = 1 \quad 4$	$\text{Na} = 2 \quad 4$
$\text{O} = 2$	$\text{O} = 1 \quad 2$

1. Copper (II) oxide is reduced by hydrogen gas



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=	=
=	=

2. Magnesium ribbon is burn completely with oxygen gas



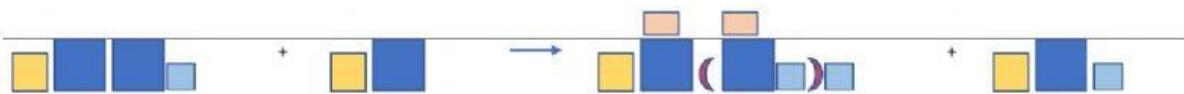
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3. Displacement of silver (II) sulphate by zinc metal



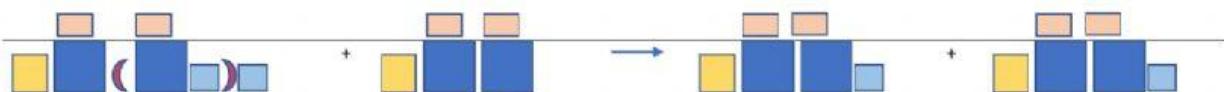
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4. iron powder is added into nitric acid



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5. lead (II) nitrate solution is added into potassium iodide solution



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