

1

Reactivity series of metals is an arrangement of metals according to the reactivity of their reactions with _____ to form _____. **TP2**

2

Fill the name of metals that involved in the reactivity series of metals towards oxygen in the box below.

Au, Ag, Hg, , Pb, Sn, Fe, , Al, Mg, Ca, , K

Reactivity of metal towards oxygen _____

The _____ the position of metal in the Reactivity Series of Metal, the more reactive the reaction of metal towards oxygen. **[TP2]**

Logam Metal	Tindak balas dengan oksigen Reaction towards oxygen	Kereaktifan logam Reactivity of metal
Kalium / Potassium (K) Natrium / Sodium (Na) Kalsium / Calcium (Ca) Magnesium / Magnesium (Mg)	Burns very brightly and vigorously	_____
Aluminium / Aluminium (Al) Zink / Zinc (Zn) Ferum / Iron (Fe) Timah / Tin (Sn) Plumbum / Lead (Pb) Kuprum / Copper (Cu)	Produce yellow flame	Reaktif Reactive
Merkuri / Mercury (Hg) Argentum / Argentum (Ag) Emas / Gold (Au)	Burning	_____