What is the difference between an aldehyde and a ketone?

- A- Aldehydes have carbonyl groups; ketones have carboxyl groups.
- B- Aldehydes have carbonyl groups at the end of a carbon chain; ketones have carbonyl groups within carbon chains.
- C- Aldehydes have carboxyl groups; ketones have carbonyl groups.

D-carbonyl; ketone

D- Aldehydes have carboxyl groups at the end of a carbon chain; ketones have carboxyl groups within carbon chains.

2.An organic compound that has a carbonyl group bonded to a hydroxyl group is known as a(n)

A- alkane
B- alcohol
C- aldehyde
D-carboxylic acid
3. What is the name of the process in which a carboxylic acid reacts with an alcohol producing an este and releasing water?
A- hydration
B- halogenation
C- hydroxylation
D- condensation
4. Which of the following is not a property of the ketones?
A- form hydrogen bonds with each other
B- somewhat soluble in water
C- solvent for waxes
D-polar molecule
5.An ester is any organic compound with a carboxyl group in which the hydrogen of the group has been replaced by a(n) group.
A- carbonyl; alkyl
B- hydroxyl; ketone
C- hydroxyl: alkyl

