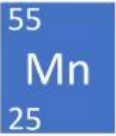
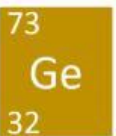


Isotope Practice:

- I. Use your periodic table to complete the chart! Remember to use the mass # given to you and not the one on the periodic table!

Name	Atomic #	Mass #	Protons	Neutrons	Electrons
Silicon-30					
Arsenic-76					
Zirconium-93					
Chlorine-35					
Lead-208					
Platinum-196					
Barium-137					
Cobalt-60					
					
					

$^{235}_{92}\text{U}$					
$^{239}_{94}\text{Pu}$					
$^{20}_{10}\text{Ne}$					

II. Use the information on your periodic table to write these elements in hyphen notation.

Na: _____

Al: _____

As: _____

Cs: _____

C: _____

Fr: _____

III. Use the information on your periodic table to write these elements in nuclear symbol notation.

Zinc: _____

Fluorine: _____

Hydrogen: _____