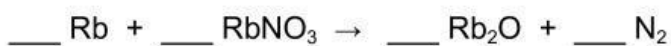
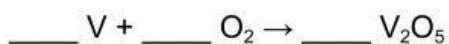


NAME _____

Station Number	Balance these equations AND upload a picture of each balanced equation you built from paper clips
1	$__\text{CuCl}_2 + __\text{Al} \rightarrow __\text{AlCl}_3 + __\text{Cu}$
2	$__\text{CO}_2 + __\text{Mg} \rightarrow __\text{MgO} + __\text{C}$
3	$__\text{N}_2 + __\text{H}_2 \rightarrow __\text{NH}_3$
4	$__\text{KClO}_3 \rightarrow __\text{O}_2 + __\text{KCl}$
5	$__\text{NaCl} + __\text{F}_2 \rightarrow __\text{NaF} + __\text{Cl}_2$
6	$__\text{O}_2 + __\text{H}_2 \rightarrow __\text{H}_2\text{O}$
7	$__\text{Pb(OH)}_2 + __\text{HCl} \rightarrow __\text{H}_2\text{O} + __\text{PbCl}_2$
8	$__\text{K}_2\text{SO}_4 + __\text{AlBr}_3 \rightarrow __\text{Al}_2(\text{SO}_4)_3 + __\text{KBr}$
9	$__\text{CH}_4 + __\text{O}_2 \rightarrow __\text{H}_2\text{O} + __\text{CO}_2$
10	$__\text{FeCl}_3 + __\text{NaOH} \rightarrow __\text{Fe(OH)}_3 + __\text{NaCl}$

Summarizer

Try to balance this equation on your own by using the table method



BALANCING EQUATIONS

