Matter



1. Choose which of the following is described.

mixture heterogeneous insulators physical change solutes and solvents pure substances properties solution chemical change homogeneous conductors

1. v	water and glass are examples of these
2. r	molecules mix
3. r	made from one type of atom or molecule
4. r	molecules are close together
5. 0	cooking an egg is an example of this
6. r	molecules can't get hot
7. H	nard or soft
8. a	substance that dissolves another
9. ł	not chocolate is an example of one
10. v	water and shampoo
11. 0	compounds are one type
12. r	molecules are free to move about
13. t	wo substances mixed together
14. r	molecules can heat up
15. r	not a pure substance
	homogeneous mixture
17. r	made from a solvent and a solute
18. €	elastic or plastic
19. r	molecules don't mix
20. ł	neterogeneous or homogeneous
21. 0	an see separate substances
22. 0	can't see separate substances
23. a	substance that is dissolved by another
24. €	elements are one type
25. f	ragile or tough
26. r	molecules move but structure stays the same
27. r	new molecules are made and matter is changed forever
28. i	ce melting is an example of this
29 r	molecules can't move