

## de Broglie and the Wave Nature of Matter

### Multiple Choice

- \_\_\_\_ 1. What happens to a high energy photon after it strikes an electron?
- decreases frequency
  - decreases wavelength
  - increases energy
  - increases momentum
- \_\_\_\_ 2. What is the deBroglie wavelength of a 125 g baseball moving at 28.0 m/s?
- $1.89 \times 10^{-34}$  m
  - $2.32 \times 10^{-33}$  m
  - $3.50 \times 10^0$  m
  - $5.28 \times 10^{33}$  m
- \_\_\_\_ 3. The deBroglie wavelength of a proton is  $5.57 \times 10^{-7}$  m. What is the speed of the proton?
- $1.19 \times 10^{-27}$  m/s
  - $3.57 \times 10^{-19}$  m/s
  - $1.28 \times 10^{-9}$  m/s
  - $7.11 \times 10^{-1}$  m/s
- \_\_\_\_ 4. What speed must a 0.20 kg ball be moving if it has a de Broglie wavelength of  $2.2 \times 10^{-34}$  m?
- 0.60 m/s
  - 15 m/s
  - 73 m/s
  - 150 m/s
- \_\_\_\_ 5. What is the de Broglie wavelength of a neutron travelling at 5.00 m/s?
- $1.58 \times 10^{-8}$  m
  - $7.91 \times 10^{-8}$  m
  - $3.96 \times 10^{-7}$  m
  - $7.92 \times 10^{-7}$  m
- \_\_\_\_ 6. What happens to the deBroglie wavelength of an electron if its momentum is doubled?
- decreases by a factor of 2
  - decreases by a factor of 4
  - increases by a factor of 2
  - increases by a factor of 4
- \_\_\_\_ 7. What is the wavelength of the matter wave associated with an electron moving at  $2.5 \times 10^7$  m/s?
- $2.9 \times 10^{-11}$  m
  - $4.7 \times 10^{-11}$  m
  - $2.9 \times 10^{-7}$  m
  - $4.7 \times 10^{-7}$  m
- \_\_\_\_ 8. What is the wavelength of a charged particle,  $q$ , which is accelerated from rest through a potential difference,  $V$ ?
- |                    |                    |
|--------------------|--------------------|
| a. $\frac{hc}{Vq}$ | c. $\frac{hcV}{q}$ |
| b. $\frac{hcq}{V}$ | d. $\frac{Vq}{hc}$ |
9. What is the deBroglie wavelength of an electron emitted with a kinetic energy of 2.4 eV?
10. Two subatomic particles with very different masses have the same de Broglie wavelength. Explain how this is possible.

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**Answer Section**

**MULTIPLE CHOICE**

- |           |          |                            |
|-----------|----------|----------------------------|
| 1. ANS: A | DIF: ii  | TOP: photon momentum       |
| 2. ANS: A | DIF: ii  | TOP: De Broglie Wavelength |
| 3. ANS: D | DIF: ii  | TOP: De Broglie Wavelength |
| 4. ANS: B | DIF: ii  | TOP: De Broglie wavelength |
| 5. ANS: B | DIF: ii  | TOP: De Broglie wavelength |
| 6. ANS: A | DIF: ii  | TOP: De Broglie Wavelength |
| 7. ANS: A | DIF: ii  | TOP: De Broglie Wavelength |
| 8. ANS: A | DIF: iii | TOP: De Broglie wavelength |

**PROBLEM**

9. ANS:  
see key
- DIF: ii      TOP: De Broglie wavelength