

### Q3 AT Molar Mass

Name \_\_\_\_\_ Date \_\_\_\_\_ Period \_\_\_\_\_

Write the formula for each compound (you thought you were done with this) and calculate the Molar Mass of each of the compounds listed below to the hundredths place. All numerical answers MUST HAVE THE CORRECT UNITS!

|    | Name                     | Formula | Molar Mass |
|----|--------------------------|---------|------------|
| 1  | cobalt(V) chloride       |         |            |
| 2  | tetraphosphorus trioxide |         |            |
| 3  | calcium selenide         |         |            |
| 4  | iron(IV) sulfate         |         |            |
| 5  | aluminum fluoride        |         |            |
| 6  | silicon trioxide         |         |            |
| 7  | copper(III) carbonate    |         |            |
| 8  | zinc phosphate           |         |            |
| 9  | lead(I) dichromate       |         |            |
| 10 | carbon hexachloride      |         |            |