



English – Class work



Look and Match. (Elige la opción correcta)

The number of protons in the nucleus of the atom. The average mass of the atoms in an element.	CARBON 6 C 12.01	Usually from a Greek or Latin word for the element or a substance containing the element. Short-hand abbreviation for the element name.
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Parts of an ATOM. (Arrastra los nombres a la definición correcta.)

Electron

Proton

Neutron

- Positively charged particle in the nucleus of the atom.
The number of protons in an atom's nucleus is the atomic number.
- Negatively charged particle surrounding the nucleus of the atom.
The number of electrons surrounding the nucleus of an atom is equal to the number of protons in the atom's nucleus.
- Particle in the nucleus that has about the same mass as a proton but has no charge. For the atoms of the first 20 elements, the number of neutrons is either equal to or slightly greater than the number of protons.