

DETERMINE THE TYPE OF PARTICLE FOR COMPOUNDS BELOW

Compound / Element <i>Sebatian / Unsur</i>	Type of particles <i>Jenis zarah</i>	Compound / Element <i>Sebatian / Unsur</i>	Type of particles <i>Jenis zarah</i>
Sodium sulphate <i>Natrium sulfat</i>		Zinc carbonate <i>Zink karbonat</i>	
Ammonium carbonate <i>Ammonium karbonat</i>		Ammonium carbonate <i>Ammonium karbonat</i>	
Magnesium nitrate <i>Magnesium nitrat</i>		Silver chloride <i>Argentum klorida</i>	
Hydrochloric acid <i>Asid hidroklorik</i>		Sulphuric acid <i>Asid sulfurik</i>	
Potassium oxide <i>Kalium oksida</i>		Copper(II) nitrate <i>Kuprum(II) nitrat</i>	
Magnesium oxide <i>Magnesium oksida</i>		Hydrogen gas <i>Gas hidrogen</i>	
Lead(II) carbonate <i>Plumbum(II) karbonat</i>		Carbon dioxide gas <i>Gas karbon dioksida</i>	
Iron(III) sulphate <i>Ferum(III) sulfat</i>		Oxygen gas <i>Gas oksigen</i>	
Magnesium chloride <i>Magnesium klorida</i>		Aluminium sulphate <i>Aluminium sulfat</i>	
Zinc sulphate <i>Zink sulfat</i>		Lead(II) chloride <i>Plumbum(II) klorida</i>	
Silver nitrate <i>Argentum nitrat</i>		Potassium iodide <i>Kalium iodida</i>	
Ammonium sulphate <i>Ammonium sulfat</i>		Copper(II) carbonate <i>Kuprum(II) karbonat</i>	
Zinc oxide <i>Zink oksida</i>		Potassium carbonate <i>Kalium karbonat</i>	
Nitric acid <i>Asid nitrik</i>		Sodium hydroxide <i>Natrium hidroksida</i>	
Ammonia gas <i>Gas ammonia</i>		Aqueous ammonia <i>Ammonia akueus</i>	
Magnesium <i>Magnesium</i>		Ammonium chloride <i>Ammonium klorida</i>	
Zinc <i>Zink</i>		Nitrogen dioxide gas <i>Gas nitrogen dioksida</i>	
Copper(II) sulphate <i>Kuprum(II) sulfat</i>		Sodium chloride <i>Natrium klorida</i>	
Iodine <i>Iodin</i>		Silver <i>Argentum</i>	
Chlorine <i>Klorin</i>		Bromine <i>Bromin</i>	