

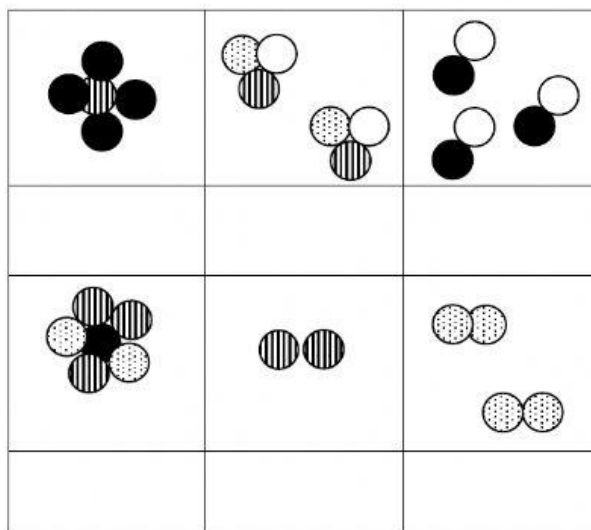
Name: \_\_\_\_\_

Date: \_\_\_\_\_

Read the chemical formulas. Write the number of molecules (coefficient), the number of elements, and number of atoms (total amount with coefficient and subscript).

Chemical Formula	Number of Molecules	Number of Elements	Number of Atoms
$2\text{H}_2\text{O}_2$			
$4\text{HCl}$			
$3\text{CH}_4$			
$\text{C}_6\text{H}_{12}\text{O}_6$			
$5\text{CaCl}_2$			
$\text{NaCl}$			
$6\text{NH}_3$			
$3\text{H}_2\text{O}$			
$2\text{Fe}_2\text{O}_3$			
$8\text{CO}$			

Look at the following reactions. Choose the formula that represents each diagram, and drag it underneath the correspondent:



$2\text{FClH}$

$\text{F}_2$

$3\text{OH}$

$\text{FO}_4$

$2\text{Cl}$

$\text{OCl}_2\text{F}_3$