

Chọn hệ số thích hợp để cân bằng các phương trình hóa học sau

- $\text{Ca} + \text{O}_2 \rightarrow \text{CaO}$
- $\text{Al} + \text{Cl}_2 \rightarrow \text{AlCl}_3$
- $\text{P} + \text{O}_2 \rightarrow \text{P}_2\text{O}_5$
- $\text{P}_2\text{O}_5 + \text{H}_2\text{O} \rightarrow \text{H}_3\text{PO}_4$
- $\text{C}_4\text{H}_{10} + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$
- $\text{Al} + \text{H}_2\text{SO}_4 \rightarrow \text{Al}_2(\text{SO}_4)_3 + \text{H}_2$
- $\text{CuCl}_2 + \text{Al} \rightarrow \text{AlCl}_3 + \text{Cu}$.
- $\text{Fe}_2\text{O}_3 + \text{CO} \rightarrow \text{Fe} + \text{CO}_2$.
- $\text{Na} + \text{O}_2 \rightarrow \text{Na}_2\text{O}$.
- $\text{HCl} + \text{CaCO}_3 \rightarrow \text{CaCl}_2 + \text{CO}_2 + \text{H}_2\text{O}$
- $\text{C}_3\text{H}_8\text{O} + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$
- $\text{Al}(\text{OH})_3 + \text{H}_2\text{SO}_4 \rightarrow \text{Al}_2(\text{SO}_4)_3 + \text{H}_2\text{O}$
- $\text{Na} + \text{H}_2\text{O} \rightarrow \text{NaOH} + \text{H}_2$
- $\text{Fe}_3\text{O}_4 + \text{H}_2 \rightarrow \text{Fe} + \text{H}_2\text{O}$.

