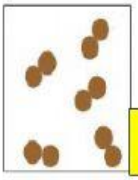
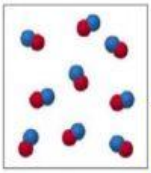
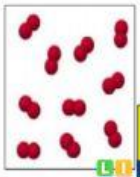
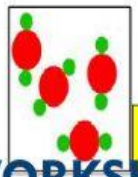
Review: Elements versus Compounds

Directions: Label each item as either **E (for element)** or **C (for compound)**.

1. Made up of 1 type of substance; is the simplest type of matter
2. Made of 2 or more substance that are chemically bonded (cannot be easily separated)
3. Sodium (Na)
4. Salt (NaCl)
5. Carbon dioxide (CO_2)
6. Carbon (C)
7. Oxygen (O_2)
8. 
9. 
10. 
11. 
12. 

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Practice: Counting Atoms in a Chemical Formula

$3C_2H_2$	
C	
H	

$KMnO_4$	
K	
Mn	
O	

$2H_2O_2$	
H	
O	

CuN_2O_6	
Cu	
N	
O	

$3Hg_2C\ell_2$	
Hg	
$C\ell$	

NaOH	
Na	
O	
H	

LIVEWORKSHEETS