

Name: _____ Date: _____

CHEMISTRY

Periodic Table of the Elements

Use the Periodic Table of the Elements to answer the questions.

1 1.0079 H Hydrogen	Atomic number — 5 10.811 — Atomic mass B — Symbol Boron — Name																2 4.0026 He Helium																				
3 6.941 Li Lithium	4 9.0122 Be Beryllium																	5 10.811 B Boron	6 12.011 C Carbon	7 14.007 N Nitrogen	8 15.999 O Oxygen	9 18.998 F Fluorine	10 20.180 Ne Neon														
11 22.990 Na Sodium	12 24.305 Mg Magnesium																	13 26.982 Al Aluminium	14 28.086 Si Silicon	15 30.974 P Phosphorus	16 32.065 S Sulfur	17 35.453 Cl Chlorine	18 39.948 Ar Argon														
19 39.098 K Potassium	20 40.078 Ca Calcium	21 44.956 Sc Scandium	22 47.867 Ti Titanium	23 50.942 V Vanadium	24 51.996 Cr Chromium	25 54.938 Mn Manganese	26 55.845 Fe Iron	27 58.933 Co Cobalt	28 58.933 Ni Nickel	29 63.546 Cu Copper	30 65.38 Zn Zinc	31 69.723 Ga Gallium	32 72.64 Ge Germanium	33 74.922 As Arsenic	34 78.96 Se Selenium	35 79.904 Br Bromine	36 83.798 Kr Krypton	37 85.468 Rb Rubidium	38 87.62 Sr Strontium	39 88.906 Y Yttrium	40 91.224 Zr Zirconium	41 92.906 Nb Niobium	42 95.96 Mo Molybdenum	43 (98) Tc Technetium	44 101.07 Ru Ruthenium	45 101.07 Rh Rhodium	46 106.42 Pd Palladium	47 107.87 Ag Silver	48 112.41 Cd Cadmium	49 114.82 In Indium	50 118.71 Sn Tin	51 121.76 Sb Antimony	52 127.60 Te Tellurium	53 126.90 I Iodine	54 131.29 Xe Xenon		
55 132.91 Cs Cesium	56 137.33 Ba Barium	La-Lu Lanthanide		72 180.95 Hf Hafnium	73 180.95 Ta Tantalum	74 183.84 W Tungsten	75 186.21 Re Rhenium	76 186.21 Os Osmium	77 188.91 Ir Iridium	78 195.08 Pt Platinum	79 196.97 Au Gold	80 200.59 Hg Mercury	81 204.38 Tl Thallium	82 207.2 Pb Lead	83 208.98 Bi Bismuth	84 (209) Po Polonium	85 (210) At Astatine	86 (222) Rn Radon	87 (223) Fr Francium	88 (226) Ra Radium	Ac-Lr Actinide		104 (261) Rf Rutherfordium	105 (262) Db Dubnium	106 (271) Sg Seaborgium	107 (272) Bh Bohrium	108 (276) Hs Hassium	109 (277) Mt Meitnerium	110 (281) Ds Darmstadtium	111 (285) Rg Roentgenium	112 (289) Cn Copernicium	113 (294) Nh Nihonium	114 (299) Fl Flerovium	115 (304) Mc Moscovium	116 (310) Lv Livermorium	117 (315) Ts Tennessine	118 (318) Og Oganesson
57 138.91 La Lanthanum	58 140.12 Ce Cerium	59 140.91 Pr Praseodymium	60 140.91 Nd Neodymium	61 (145) Pm Promethium	62 150.36 Sm Samarium	63 151.96 Eu Europium	64 157.25 Gd Gadolinium	65 158.93 Tb Terbium	66 162.50 Dy Dysprosium	67 164.93 Ho Holmium	68 167.26 Er Erbium	69 168.93 Tm Thulium	70 173.05 Yb Ytterbium	71 174.97 Lu Lutetium																							
89 (227) Ac Actinium	90 232.04 Th Thorium	91 231.04 Pa Protactinium	92 238.03 U Uranium	93 (237) Np Neptunium	94 (244) Pu Plutonium	95 (243) Am Americium	96 (247) Cm Curium	97 (247) Bk Berkelium	98 (251) Cf Californium	99 (252) Es Einsteinium	100 (257) Fm Fermium	101 (258) Md Mendelevium	102 (259) No Nobelium	103 (267) Lr Lawrencium																							

Part 1: Atomic Number. On the line next to the element, write the atomic number of the element. Your answer should be 1, 2, or 3 digits. No letters.

- | | |
|---|---|
| 1. <input style="width: 50px; height: 25px;" type="text"/> Hydrogen
2. <input style="width: 50px; height: 25px;" type="text"/> Oxygen
3. <input style="width: 50px; height: 25px;" type="text"/> Magnesium
4. <input style="width: 50px; height: 25px;" type="text"/> Vanadium
5. <input style="width: 50px; height: 25px;" type="text"/> Krypton
6. <input style="width: 50px; height: 25px;" type="text"/> Ruthenium | 7. <input style="width: 50px; height: 25px;" type="text"/> Cerium
8. <input style="width: 50px; height: 25px;" type="text"/> Iridium
9. <input style="width: 50px; height: 25px;" type="text"/> Bismuth
10. <input style="width: 50px; height: 25px;" type="text"/> Uranium
11. <input style="width: 50px; height: 25px;" type="text"/> Fermium
12. <input style="width: 50px; height: 25px;" type="text"/> Darmstadtium |
|---|---|

Part 2: Which element. On the line next to the element, write the chemical symbol of the element. Your answer should be 1 or 2 letters. If one letter, the letter must be capitalized. If two letters, the first letter must be capitalized and the second lower case.

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|-----|----------------------|------------|-----|----------------------|------------|
| 13. | <input type="text"/> | Helium | 19. | <input type="text"/> | Molybdenum |
| 14. | <input type="text"/> | Nitrogen | 20. | <input type="text"/> | Silver |
| 15. | <input type="text"/> | Sodium | 21. | <input type="text"/> | Europium |
| 16. | <input type="text"/> | Phosphorus | 22. | <input type="text"/> | Tungsten |
| 17. | <input type="text"/> | Copper | 23. | <input type="text"/> | Lead |
| 18. | <input type="text"/> | Strontium | 24. | <input type="text"/> | Berkelium |

Part 3: Which Type of Element. On the line next to the element, type the letter of the “class of element”. The choices are below. Only type the letter. Letters may be used multiple times. Examples of elements (chemical symbol) that fall into that class of element have been provided to assist you.

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|--------------------------------|------------------------------|
| A Alkali metal (Li) | F Actinide series metal (Pu) |
| B Alkali earth metal (Ca) | G Metalloid element (Si) |
| C Transition metal (Fe) | H Nonmetal element (O) |
| D Post-transition metal (Al) | I Halogen element (Cl) |
| E Lanthanide series metal (Ce) | J Noble Gas element (Kr) |

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|-----|----------------------|------------|-----|----------------------|------------|
| 25. | <input type="text"/> | Magnesium | 31. | <input type="text"/> | Osmium |
| 26. | <input type="text"/> | Phosphorus | 32. | <input type="text"/> | Dysprosium |
| 27. | <input type="text"/> | Titanium | 33. | <input type="text"/> | Cesium |
| 28. | <input type="text"/> | Arsenic | 34. | <input type="text"/> | Lead |
| 29. | <input type="text"/> | Radon | 35. | <input type="text"/> | Platinum |
| 30. | <input type="text"/> | Uranium | 36. | <input type="text"/> | Fluorine |