

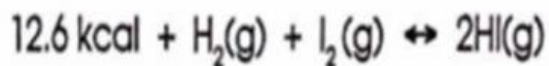
Le Chatelier's Principle

For this assignment use the following (ALL LOWERCASE):

r for right
l for left
i for increase
d for decrease
s for same

Is the equation below exothermic or endothermic?
Spell it out, lowercase, next to the equation.

When you are done click finish, it will give you a grade. Upload your screenshot to the assignment.



Stress	Equilibrium Shift	[H ₂]	[I ₂]	[HI]	K
1. Add H ₂	right	_____	decreases	increases	remains the same
2. Add I ₂			_____		
3. Add HI				_____	
4. Remove H ₂		_____			
5. Remove I ₂			_____		
6. Remove HI				_____	
7. Increase Temperature					
8. Decrease Temperature					
9. Increase Pressure					
10. Decrease Pressure					