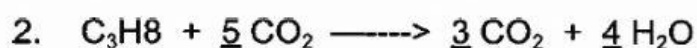




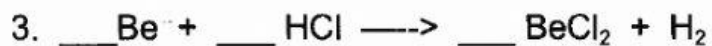
a. You begin this reaction with 25 g of lithium hydroxide. What is your theoretical yield of lithium chloride?

b. You actually produced 8 g of lithium hydroxide. What is your percent yield?



a. If you, a 10th grade Chemistry student, start with 5 g of C_3H_8 , what will your theoretical yield of water be?

b. If you got a % yield of 75%, how many g of water did you make?



Your theoretical yield of Beryllium chloride was 12.7 g. If your actual yield is 9.5 g, what is your percent yield?



a. What will your theoretical yield of Sodium oxide be if you start with 20 g of calcium oxide?

b. Your actual yield of CaO is 7 g, what is your percent yield?