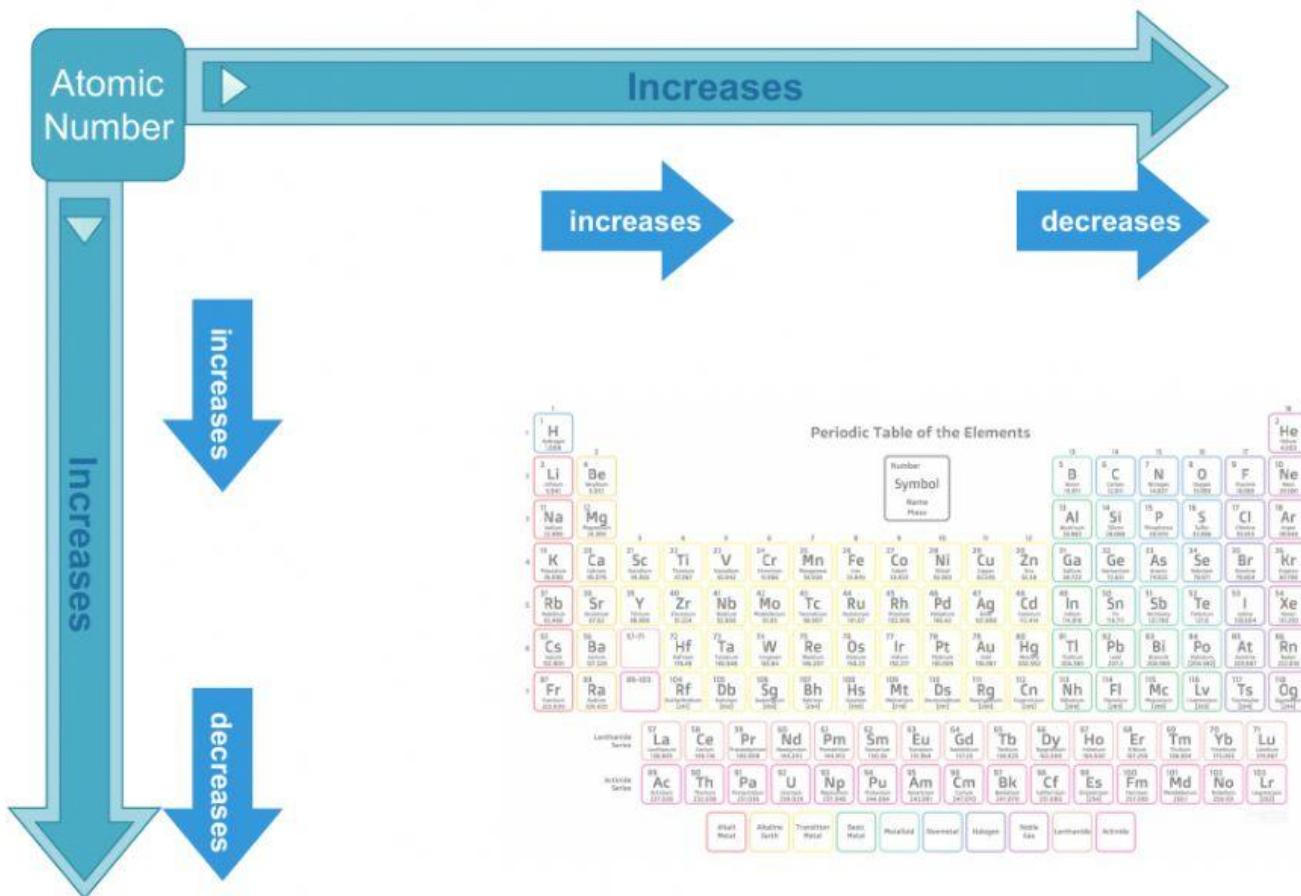


## Graduation of elements properties in modern periodic table



By increasing atomic number in group ..... increase due to increasing number of energy levels.

By increasing atomic number in period atomic size ..... due to increasing attraction force between nucleus and outer energy level's electrons.

electronegativity increases by increasing atomic number in ..... and decreases by increasing atomic number in .....

strongest metals found at ..... side of modern periodic table, while strongest non metals found at the ..... side before inert gas

the metalloid is found between ..... element and ..... element

The measuring unit of atomic radius is .....

metal oxide is ..... oxide, while non-metal oxide is ..... oxide