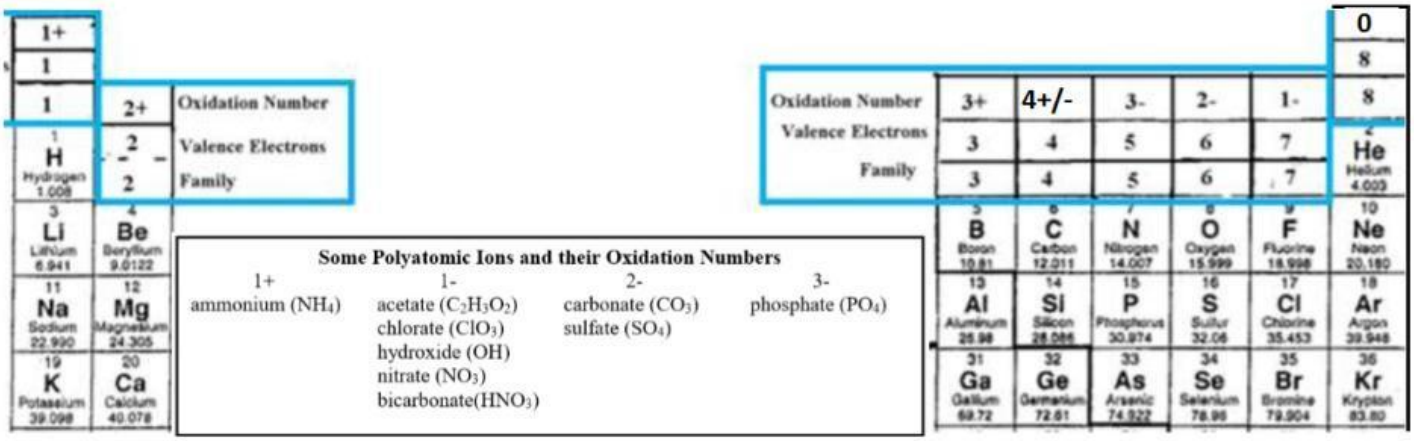


	Symbols & Oxidation #	Formula	Name of Compound
15. Magnesium and chlorine			
16. Potassium and acetate			
17. Aluminum and sulfate			
18. Lithium and nitrogen			

19. An ion is an atom or group of atoms that has become electrically _____

20. When an atom loses an electron its charge is (***positive or negative***)

21. An ionic bond is the attraction between (***opposites, positive, neutral, or negative***) ions.

22. Ionic compounds are electrically (***charged, positive, neutral, or negative***).

23. The sum of the charges for an ionic compound is _____.

The two answers must be in the right order.

24. An ionic compound is the result of the bonding of a (***non-metal, metalloid, metal, noble gas***) with a (***non-metal, metalloid, metal***).

LIVEWORKSHEETS