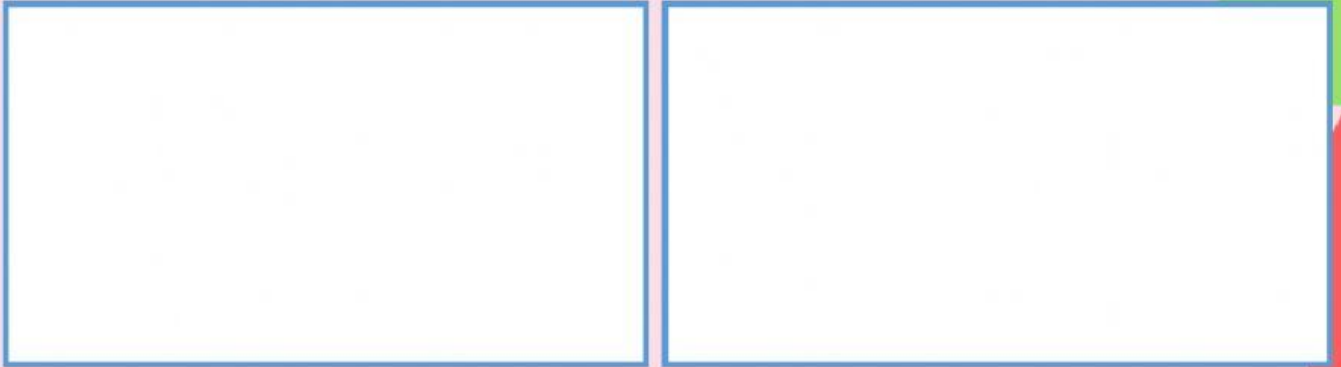


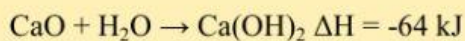
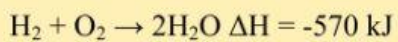
# TERMOKIMIA

Perhatikan video berikut ini!



Kerjakan latihan soal berikut ini!

1. Diketahui data reaksi sebagai berikut



Maka entalpi pembentukan  $\text{Ca(OH)}_2$  adalah...

- A. – 984 kJ
  - B. – 856 kJ
  - C. – 1161 kJ
  - D. – 1966 kJ
  - E. – 1904 kJ
- 2.

Entalpi

$$H = E \times W$$

Perubahan Entalpi

$$H_p > H_r$$

Reaksi Endoterm

$$H_p < H_r$$

Reaksi Eksoterm

$$\Delta H = H_p - H_r$$