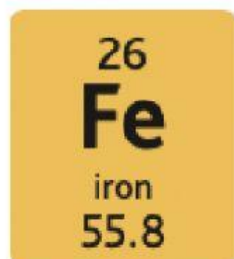
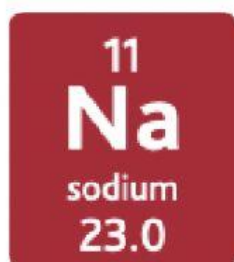
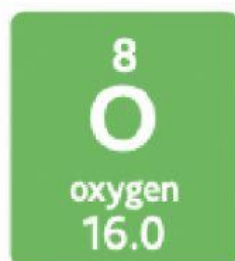


In the space below, write the unabbreviated electron configurations of the following elements:

Drag the correct electron configuration to the correct element below.



$1s^2 2s^2 2p^6 3s^1$

$1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^{10} 4p^5$

$1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^6$

$1s^2 2s^2 2p^4$

Drag the correct element to the correct electron configuration

$1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^{10} 4p^6 5s^1$

$1s^2 2s^2 2p^6 3s^2 3p^4$

$[\text{Kr}] 5s^2 4d^{10} 5p^3$

