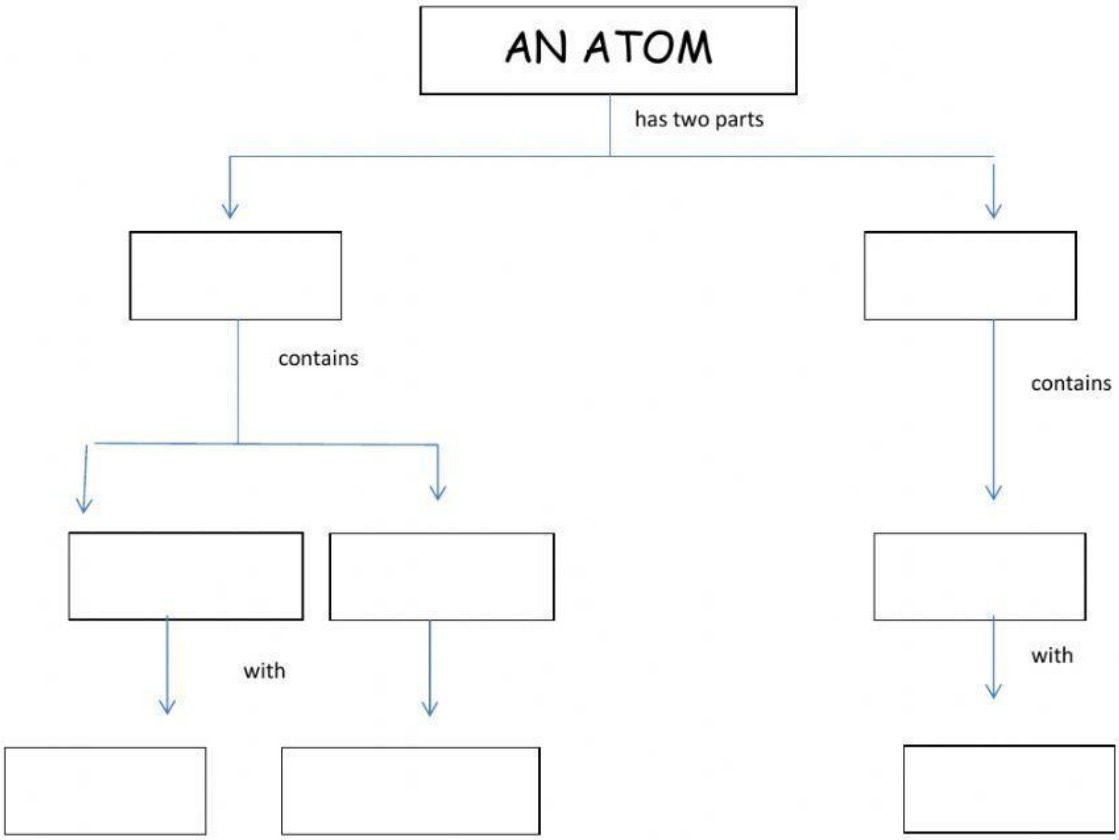
# STRUCTURE OF THE ATOM

1) Complete the boxes with the characteristics of the atom using these words: NEUTRONS – NEGATIVE CHARGE- SHELLS – PROTONS – ELECTRONS – POSITIVE CHARGE – NO CHARGE - NUCLEUS




2) Decide if these characteristics belong to the PROTONS, NEUTRONS or ELECTRONS

a- They have no electrical charge:

b- They are negatively charged particles:

c- They are located in the nucleus:

d- They are positively charged particles:

e- They are the smallest subatomic particles:

f- They don't move:

g- They spin around the nucleus:

h- They are located in the shells:

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3) Correct the mistakes (all the statements are wrong but one)

a- Protons and electrons are in the nucleus.

b- Electrons don't move.

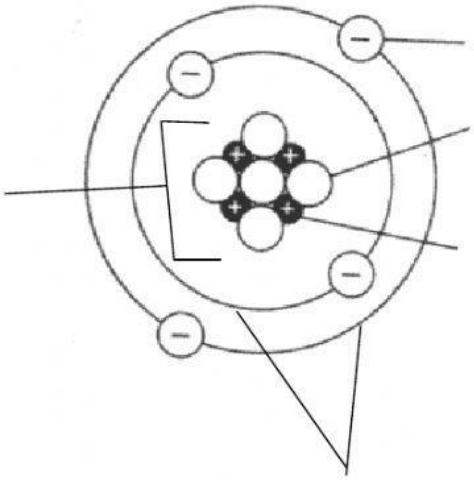
c- Protons and electrons tend to repel each other.

d- All the atoms have the same number of protons and electrons.

e- There are only two subatomic particles.

f- Only solids are made of atoms:

4) Label the parts of the atom



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